DEVELOPMENT OF BIOACTIVE CELLULOSE SULFATES FOR BIOMEDICAL APPLICATIONS

Thomas Groth^{1,2}, Christian Willems¹, Kai Zhang³, Steffen Fischer⁴

Cellulose is the most abundant biomolecule on earth. Chemical derivatives of cellulose have found multitude of uses in industrial and biotechnological applications. Cellulose sulfates (CS) represent a class of water-soluble derivatives that have been employed in industrial application, but not yet in medicine. Here derivatives with different degree of sulfation of anhydroglucose unit (AGU) of cellulose have been studied toward anticoagulant effects and modulating effects of growth factors with heparin-binding domains like fibroblast growth factor 2 (FGF-2). The results show that CS of higher sulfation degree have an anti-coagulant activity comparable to that of heparin with cooperative action to anti-thrombin III that inhibits thrombin and Factor Xa activity making CS interesting for anticoagulant coating of blood-contacting medical devices. Furthermore, the studies show that CS with comparable sulfation degree to heparin have a promoting activity on the mitogenic effect of FGF-2 shown in cell culture studies that indicate their application as coatings of implant materials or component of tissue engineering scaffolds in the area of traumatology and regenerative medicine. *Acta Medica Medianae 2020;59(3):56-67.*

Key words: cellulose sulfates, sulfation degree, anticoagulation, thrombin, growth factors, FGF-2

¹Department of Biomedical Materials, Institute of Pharmacy, Martin Luther University Halle-Wittenberg, Halle (Saale) ²Interdisciplinary Center of Materials Science, Martin Luther University Halle-Wittenberg, Halle (Saale), Germany ³Wood Technology and Wood Chemistry, Georg-August-University of Goettingen, Göttingen, Germany ⁴Institute of Plant and Wood Chemistry, Technical University Dresden, Germany

Contact: Thomas Groth Heinrich-Damerow-Strasse 4, 06120 Halle, Saale, Germany E-mail: thomas.groth@pharmazie.uni-halle.de

Introduction

Glycans have diverse functions in the organism. They are relevant not only for maintaining the normal functions of cells and tissue, they are also important in wound healing or pathological processes, like bacterial or viral infections, growth and metastasis of tumors (1). An important subgroup of glycans is glycosaminoglycans (GAGs), which comprise specific dimeric repeating units that usually consist of uronic acid connected to a N-acetylglucosamine or N-acetylgalactosamine sugar. Important 56 droitin sulfate, heparan sulfate and heparin. It is interesting to note that the degree of sulfation is highest in heparin, followed by heparan sulfate and other sulfated GAGs, while hyaluronic acid is the only non-sulfated GAG (1, 2). Heparan sulfate and heparin are structurally quite similar, but differ mainly in their degree of sulfation, molecular weight and occurrence in the organism (3). Because of its high number of sulfate and carboxyl groups, heparin is the glycosaminoglycan with the highest negative charge density (Figure 1). Heparin is composed of disaccharide-subunits that are either D-glucuronic acid (10 %) bound by a β -(1 \rightarrow 4) glycosidic bond or L-iduronic acid (90%) bound by an $a-(1\rightarrow 4)$ glycosidic bond to N-Acetyl-D-glucosamine. The typical structure is usually the trisulfated disaccharide. Sulfate groups can be located at the 2-O position of the iduronic or glucuronic acid, as well as the 3-O and 6-O position of the glucosamine. The amino group of the glucosamine on the other hand can be either substituted by an acetyl or sulfate group. The average molecular weight of a heparin molecule is about 15 kDa (4). By comparison, heparan sulfate has a higher average molecular weight (around 30 kDa) and a lower degree of sulfation. Heparan sulfate chains frequently contain domains with long sequences of either high or low sulfation degree. Heparin shows such inhomogeneities in its degree of sulfation as well, but to a lesser extent in comparison to heparan sulfate (5). The synthesis of heparin hap-

representatives of GAGs are hyaluronic acid, chon-

pens mainly in mast cells and basophil granulocytes. Heparin shows a multitude of functions, such as the binding of histamine and the control of the activity of different proteases (6). Heparin that has been released also inhibits the interactions of blood platelets with collagen and binding of von Willebrand factors to them. In addition, it can also bind low density lipoprotein (LPL). Such bound constructs can subsequently be taken up and degraded by macrophages (8). However, the well-known property of heparin is its ability to bind to anti-thrombin III (AT III), which enhances AT III affinity to thrombin and other activated clotting factors of the coagulation cascade, which is the basis of its use as anticoagulant (6).



Cellulose sulfate

Figure 1. Chemical structure of heparin and cellulose sulfate

On the other hand, heparan sulfate proteoglycans (HSPGs) are localised on the surface of many cells and represent components of the extracellular matrix. The most known representatives of cell membrane localised HSPGs are syndecans and glypicans3. For example, syndecan-1 und -3 carry chondroitin sulfate proximal to the cell membrane, while heparan sulfate (HS) is found at the distal part of proteoglycans. Conversely, syndecan-2 and -4 are decorated exclusively with heparan sulfates.

Because of their cytoplasmic domain, syndecans can transmit signals from the extracellular area to the inner part of the cell. This includes the binding of ligands to the HS chains, followed by oligomerisation of syndecans, which triggers the activation of signalling proteins like kinases in the cytoplasm (1). Many functions of the HSPGs are also related to the regulation of the activity of chemokines, growth factors glycoproteins of the extracellular matrix (ECM) in which they can function as co-receptors. By presenting growth factors as co-receptors for receptor-tyrosine kinases, they can contribute to signal transduction processes (9). The HS chains are directly involved in the formation of the receptorligand complex, as described for the fibroblast growth factor FGF-2, and can influence the mitogenic effect of cytokines (10). HSPG can also store growth factors outside the cell as component of ECM and essentially function as a reservoir. The release can be caused either through a change in the degree of sulfation through locally expressed sulfatases (11), through proteolysis of the protein backbone or through the HS chain fission caused by heparanases. The binding of heparin or HS to regulatory proteins occurs through heparin binding sites, which are generally located at the outside of proteins. Those are mostly rift-like domains with a high amount of positively charged amino acids like lysine or arginine (4).

The inhibition of blood coagulation through heparin is an effect that has been used in hospitals for a long time (4). Heparin is also used to make the surface of biomaterials like catheters, tube systems or dialysis membranes more compatible to blood (12). This can reduce the need for a systemic application of heparin, since that can cause side effects like the aggregation of thrombocytes, an increased tendency for bleeding or a retarded healing process of bone tissue and an increased risk for osteoporosis (13, 14). Besides these well-known risks, there are further disadvantages in the use of natural GAGs since the extraction of them out of animal tissue is time-consuming and carries further risk of infections and immunological reactions during their clinical application. The bioactivity of heparin is additionally very dependent on its biological origin (species, organs), which is related to differences in degree of sulfation and substitution pattern (4). Lastly, the use of natural GAGs in the clinical field opens the possibility to contaminate heparin with highly sulfated chondroitin sulfate falsely claimed with criminal intent as pure heparin, which was fatal for some patients in US (15). Because of these challenges, it would be highly desirable to synthesize biocompatible polymers, which can replace heparin showing less variance in their biological activity with no risk of transmission of diseases and hence better safety.

Cellulose is one of the most abundant polysaccharides that exist in nature. It is a main component of the cell wall of plants, shows a high molecular weight, but is non-soluble in water and most organic solvents. In comparison to other natural polysaccharides like hemicellulose or pectin, cellulose is a non-branched polymer (16). It is composed of D-glucose units, which are connected via β- $(1\rightarrow 4)$ glycosidic bonds. The hydroxyl groups at the C2, C3 and C6 atoms of the anhydroglucose unit (AGU) can be chemically functionalized to synthesize many widespread polymers such as e.g. carboxymethyl cellulose, which are used in the paper industry, food technology and partly in medical applications (17). The sulfation of cellulose has a long tradition and leads to water soluble products with many different application possibilities (18). The main structure of cellulose sulfate (CS) is shown in Figure 1B. It is obvious that sulfation of cellulose leads to derivatives that have similarities to the highly N-acetyl glucosamine unit of heparin. Hence, it seems to be reasonable to assume that CS of higher sulfation degree might be also effective in inhibition of blood coagulation and in supporting the activity of growth factors that possess heparinbinding domains. This article presents the effect of CSs on coagulation and the mitogenic activity of the fibroblast growth factor 2 (FGF-2) showing that sulfation degree has an effect on both phenomena.

Materials and methods

Synthesis of cellulose derivatives

We have already described synthesis and chemical analysis of cellulose derivatives in more detail (28, 29) and therefore will not be described here in detail. The CSs were named later in the result section according to the degree of substitution with sulfate (DS) as CS X.

Analytical methods

The DS of the cellulose derivatives obtained by different sulfation methods was characterized by

elemental analysis and 13 C-NMR spectroscopy. The substituent distribution within the AGU was assessed from the 13 C-NMR spectrum of the cellulose derivatives dissolved in D₂O by integrating the signal areas and comparing those of the substituted position to those of the appropriate non-substituted one.

Study on anticoagulant activity of cellulose sulfates

Collection and preparation of blood

Blood was drawn from healthy human volunteers, who had no medication for at least 10 days. Blood was anticoagulated with sodium citrate (3.8 g/100 ml). The blood was centrifuged at 2000g for 20 min. The supernatant cell free plasma was separated. Plasma samples from 10 different donors were pooled, aliquoted and snap-frozen at -80 °C. For experimental work, plasma was thawed at 37 °C and used within 2 h.

Measurement of clotting times

Thrombin time (TT) was measured using thrombin (Behring Werke, Germany). Partial thromboplastin time (PTT) was estimated using a commercial test kit (Boehringer Mannheim, Germany). Measurements were carried out with a coagulometer KC 4A (Amelung, Germany). Cellulose derivatives were dissolved in TRIS buffer, pH 7.4. 100 µL pooled plasma were mixed with 50 µL cellulose derivative solution and incubated for 1 min (TT) or 3 min (PTT), respectively. TT was measured after the addition of 100 µL thrombin solution (0.3 IU/mL). PTT was estimated after the addition of 100 µL kaolincephalic solution, followed by the addition of 100 μ L 25 mM CaCl₂ solution. After the addition of activator, the time needed for clotting was measured. If samples did not clot within 10 min, it was observed that no clotting occurred afterwards. Therefore, measurements were stopped after 10 min and those samples denoted as non-clottable (n.c.). To still obtain visible data points in the graphs, these values were set at a clotting time of 600 s. However, it should be kept in mind that these data represent conditions under which the plasma did not clot at all.

Inactivation of thrombin and factor Xa

The anticoagulant potential of CSs was tested in addition by their ability to support the inactivation of thrombin (F IIa) and factor Xa (F Xa) in the presence of antithrombin III (AT III). This was possible by the development of amidolytic assays for F IIa and F Xa in separate investigations. Cellulose derivatives or reference substances were dissolved in 50 mM Tris-HCl,175 mM NaCl, 10 mM EDTA, and 0.5 mg/mL human serum albumin (24).

The thrombin assay was carried out mixing 50 μ L AT III (activity 0.265 pkat/mL) with 200 μ L F IIa (activity 0.53 nkat/mL), and 50 μ L of the test substance. After 5 min incubation at 37 °C, 200 μ L chromogenic substrate S-2238 (0.22 mM) was added and the mixture was incubated for 2 min. The con-

version of the chromogenic substrate was stopped by the addition of 200 μ L acetic acid (20% v/v). The optical density (OD) was measured at 405 nm in 96 well plates with a plate reader (Anthos 2001, Austria). A standard curve was obtained under identical conditions for thrombin activities from 0 up to 1.053 nkat/mL and used for the calculation of residual thrombin activity from the measured OD.

The F Xa assay was performed using 200 μ L of F Xa solution (activity 1.06 nkat/mL), 50 μ L AT III solution (activity 0.265 pkat/mL), 50 μ L test solution and 200 μ L chromogenic substrate S-2222 (0.22 mM). The experiment was carried out in the same manner as the thrombin assay. Residual F Xa activities were calculated from a standard curve. Thrombin, factor Xa, AT III, and the chromogenic substrates S-2238, and S-2222 were supplied by Chromogenix, Sweden.

Studies on mitogenic activity of cellulose sulfates

Estimation of binding growth factor FGF-2 to cellulose sulfates

The binding affinity of the synthesized CSs to the growth factor FGF-2 (bFGF) was performed with a competition assay using heparin agarose beads (Fluka, Biochemica). 25 ng of b-FGF obtained from InVitrogen (Karlsruhe, Germany) were mixed with heparin agarose beads and PBS and agitated for 30 min at 200 rpm at RT to allow the binding of growth factors to the beads. The unbound growth factor was removed by washing with PBS twice. For the release of the growth factor from the beads, CSs or heparin (control) were added to the mixture and agitated for 30 min at 200 rpm at RT. After centrifugation, the supernatants with the polysaccharides and the released FGF-2 were applied to a cellulose nitrate membrane in a slot-blot apparatus. A primary antibody against FGF-2 (Sigma, Germany) and a horseradish peroxidase labelled secondary antibody (Dianova, Hamburg, Germany) were applied to the membrane to label bound growth factors. Detection was performed with ECL plus chemiluminescence kit and a CCD camera (Raytest, Diana 2). The quantification of the signals was done by ImageJ.

Cell culture

3T3-L1 fibroblast cells obtained from ATCC (Manassas, USA) were cultured in flasks (75 cm², Greiner bio-one, Frickenhausen, Germany) in Dulbecco's modified Eagle medium (DMEM, Biochrom AG, Berlin, Germany) supplemented with 10% fetal bovine serum (FBS, Biochrom AG) and 1% penicil-lin-streptomycin-fungizone (PSF, Promocell, Heidelberg, Germany) in a 37 °C humidified atmosphere of 5% CO₂ and 95% air. Cells were harvested by treatment with trypsin/EDTA (Biochrom AG). Trypsinization was stopped by the addition of FBS and cells were washed twice with DMEM.

Investigation of mitogenic effects of cellulose derivatives on 3T3-L1 fibroblasts

3T3-L1 fibroblast cells were seeded at a density of 10.000 cells/well in black 96 well plates (Greiner bio-one) in DMEM supplemented with 10% FBS and 1% penicillin-streptomycin-fungizone and cultured for 24 h. After washing the plates with DMEM only, the cellulose derivatives or heparin were applied to the cells in DMEM without FBS at a concentration range of 1 µg to 1000 µg/ml for 48 h in the presence or absence of 10 ng/ml FGF2. The proliferation was measured based on the DNA content using the Quant -iT[™] PicoGreen dsDNA quantification assay (Invitrogen, Karlsruhe, Germany). The fluorescent intensity was measured with an excitation wavelength of 485 nm and an emission wavelength of 520 nm by the plate reader Fluostar Optima. The proliferation was expressed as a ratio to the control wells with 10 ng/ml FGF2. All experiments were carried out with six wells per sample and dilution from which means and standard deviation were calculated.

Results and discussion

Studies on anticoagulant activity of cellulose sulfates

The inhibition of blood coagulation through cellulose sulfates

The sulfation of partially substituted cellulose acetates has been described elsewhere in more detail (19). The total content of sulfur was determined through elemental analysis, while the distribution of the substituents was determined through quantitative ¹³C-NMR spectroscopy. The synthesized CSs are listed in Table 1 and sorted by degree of sulfation or substitution (DS) and the distribution of the substituents. Figure 1 shows the typical structure of sulfated cellulloses. It is visible that the DS reached from low of about 0.25 to relatively high of 1.35, which is lower than that of heparin. The latter can have two sulfation sites at the N-acetylglucosamine unit with sulfation at C2, C3 and C6 position, while the uronic acid may be substituted at C2 position with a sulfate group (see Figure 1). Hence, the overall DS can have the maximum around 2.0.

The effect of CS on blood coagulation was determined using citrate plasma and commercial test kits for measurements of thrombin time (TT) and partial thromboplastin time (PTT). Figures 2A and B show the results of the TT and PTT coaqulation time measurements. It can be seen that an increase in the degree of sulfation of CS leads to an increase in the coagulation times. In addition, the results imply that the inhibition of coagulation increases in the case of TT and PTT, if the degree of sulfation is increased in C2 position, which can be shown by comparing the samples CS1.33 and CS1.35 that have almost the same degree of sulfation, but at different substitution site. (Table 1). This is especially obvious in the case of the determination of TT, since an increase of the DS at C6

and a decrease at C2 position shows a shorter time of coagulation in comparison to a sample that has a similar DS but a higher sulfation in C2 position. At a CS concentration of 25 μ g/mL with a DS \geq 0.95 for TT measurements and a DS \geq 1.15 for PTT

measurements, the blood clotting was completely inhibited, which demonstrates the potential of these CSs to be used as anticoagulant for the modification of blood-contacting materials surfaces like membranes for haemodialysis, blood linings, etc.

Table 1. Degree of sulfation (DStotal) and distribution of the sulfate groups in the derivatives in the determiantion of coagulation inhibition

Cellulose sulfate (CS)	DS _{total} Elemental analysis	DS _{total} ¹³ C-NMR	Substitution pattern of sulfates*		
			C2	C3	C6
CS 0.26	0.35	0.25	0.17	0.08	0
CS 0.95	0.80	0.95	0.55	0.20	0.20
CS 1.14	1.10	1.14	0.74	0.09	0.31
CS 1.33	1.40	1.33	0.76	0.10	0.47
CS 1.35	1.07	1.35	0.67	0.33	0.35

* DS values of the sulfate groups at the C2, C3 and C6 position were determined via 13C-NMR spectroscopy



Figure 2. Thrombin time (A) and partial thromboplastin time (B) of citrate in the presence of cellulose sulfates (concentration range: 0.025 µg/mL to 2.5 mg/mL). (♦) – CS 0.26; (●) – CS 0.95; (□) - CS 1.14; (○) - CS 1.33; (■) - CS 1.35. Asterisks in the figures show significant deviation (p <= 0.05).

Next, we were interested to find out more about the underlying mechanism of inhibition of coagulation regarding the activity of AT III in the presence of CSs toward coagulation F IIa (thrombin) and F Xa inhibition. It is known that for the inhibition of F Xa shorter heparin sequences with specific patterns of sulfation are relevant because of the formation of a binary complex between AT III and heparin that inhibits F Xa. For F IIa inhibition on the other hand, longer heparin sequences play a role because they form a ternary complex in which heparin forms a guasi-catalytic surface for the interaction of AT III and F IIa (3, 4). In analysing in which way the regioselective derivatisation of cellu-lose with sulfate groups inhibits the blood clotting, some clues were already present that the inhibition of thrombin was stronger, as seen in the TT mea-surements in comparison to PTT measurements. Because of this, tests with F IIa and F Xa were performed in the presence of AT III, selecting CSs with high and low degree of sulfation. For this, single factor tests with chromogenic substances for the specific serine proteinases F IIa and F Xa were de-veloped (19). Figure 3A shows that CS 0.26 showed no inhibiting effect against thrombin, since the re-sidual activity of thrombin was roughly 100% even at CSs concentrations of 2.5 mg/mL. Conversely, sulfated cellulose with a DSs 1.33 (CS 1.33) demon-strated an inhibition of about 40% even at concen-trations as low as 2.5 µg/mL, which points also to the importance of the overall sulfation degree. Also, the sites of sulfation seemed to be important since compared to CS 1.33, the derivative CS1.35 showed an inhibition of about 10%, only. This surprising result can be partially explained with the different substitution pattern of both samples. As already

shown in Table 1, the sample CS 1.35 has a higher degree of substitution on position C3, which leads to a lower substitution on position C2 and C6. Although C3 sulfation is critical for an anti-thrombogenic effect (especially for heparin) (20, 21), the case seems different for cellulose sulfates because of their different structure and the β -(1 \rightarrow 4) glycosidic bond. Here it seems that a higher sulfation in C2 and C6 position is more beneficial in terms of the inhibition of thrombin. The existence of sulfate groups on the C2 position of the iduronic acid of the heparin is important for the chain conformation and results in a high affinity for anti-thrombin III (20, 21), which might support the anticoagulant activity of the cellulose-2-6 sulfates towards thrombin, too. However, the results of F Xa assav look different (Figure 3B). Here both derivatives with a high DS show a significant inhibition of F Xa at higher concentration ranges starting at 50 µg/mL compared to the low sulfated CS0.26. In summary, these investigation show that CSs with a high DS in position C2 and C6 show an effectiveness analogous to heparin regarding the inhibition of blood coagulation, especially towards the inhibition of thrombin.



Figure 3 A and 3 B. Residual activity of thrombin (F IIa, A) and factor Xa (F Xa, B) after the addition of chosen cellulose sulfates with different degrees of derivatisation. (♦) - CS 0.26; (○) - CS 1.33; (■) - CS 1.35. Asterisks in the figures show significant deviation (p <= 0.05).</p>

The influence of the degree of derivatisation and regioselectivity of cellulose sulfates on the binding and activity of the fibroblast growth factor FGF-2

Here, a range of CSs were synthesized through acetosulfation or direct sulfation by which a wide range of derivatisation degrees was achieved. Table 2 gives an overview on the synthesized CSs. The distinctive feature of this sulfation procedure is that there was no detectable derivatisation at C3 position of the AGU. The details of this sulfation procedure and characterization of cellulose derivatives can be found in the previous work published by Peschel et al (22).

Cellulose	DS _s (¹³ C-NMR)*				
sulfate (CS)	C6	C2	C3	DS _{total}	
CS 0.39	0.36	0.03	0	0.39	
CS 0.58	0.52	0.06	0	0.58	
CS 0.66	0.60	0.06	0	0.66	
CS 0.92	0.77	0.15	0	0.92	
CS 1.57	1.0	0.57	0	1.57	
CS 1.69	1.0	0.69	0	1.69	
CS 1.80	1.0	0.80	n.d.	1.80	
CS 1.94	1.0	0.94	n.d.	1.94	

Table 2. Degree of sulfation (DStotal) and distribution of the sulfate groups in the derivatives in the determiantion of the activity of the growth factor FGF-2

* DS-values of sulfate groups at the C2- C3 and C6 position were determined 13C-NMR spectroscopy. n.d. – not determined

Binding of the fibroblast growth factor FGF-2 to cellulose sulfates

Growth factors like FGF-2 are presented to their corresponding receptor tyrosine kinases on the cell surface through proteoglycans like syndecan, which is decorated with heparan sulfate side chains followed by the activation of cellular kinases that induce cell proliferation (4, 5, 9). Because of this, the binding affinity of FGF-2 to CSs was determined in a competitive approach. Figure 4 shows the DS of CSs, whereas the light grey and dark grey bars illustrate the derivatization in C2 and C6 position and the observed binding of FGF-2 in comparison to heparin as a control (100%). It can be seen that in comparison to heparin, CSs with a DS \leq 0.92 allow no significant binding of FGF-2. Only for samples with a DS \geq 1.57, an increase in the growth factor binding correlating with an increase in DSS up to 60% compared to heparin could be verified. An increase of ca. 40% (p < 0.05) was found between the samples of CS 1.94 and CS 1.57.



Figure 4. Binding of FGF-2 to sulfated cellulose in dependence of the degree of sulfation in position C2 (light grey columns) and C6 (dark grey columns) compared to the binding to heparin (100%). The binding of FGF-2 is plotted against the degree of sulfation. Values represent the average ± standard deviation (n = 4).

The binding affinity is an important indicator for the biological efficacy of CSs since the binding of growth factors through GAG's often correlates with the in vitro measured biological activity (23). Here a significant binding of CSs to FGF-2 could only be verified for samples with a DS \geq 1.00 in C6 position of the AGU, which apparently correlates with an increasing DS in C2 position. Research of other authors could show that none of the hydrogen bonds that develop during the binding of heparin to FGF-2 are realized with the 6-0 (C6) sulfate groups, while the sulfation of the 2-0 (C2) or the 2-N (C2) position is of some importance for this bond24. This might be the case for samples with a DS > 1.5 that express also stronger binding of FGF-2.

Determination of the effect of cellulose sulfates on FGF-2 induced proliferation of mouse fibroblasts

To determine the influence of FGF-2 induced proliferation through CSs, a cell culture comprised of

embryonic mouse fibroblasts (cell line 3T3-L1) was used, analogous to works of other authors (25). For this, fibroblasts were incubated in a culture medium without serum adding 10 ng/mL FGF-2 and CSs for 48 hours. An additional incubation was done with heparin as a control group and cell growth was determined with a Pico-Green DNA quantification assay.

Mitogenic activity of cellulose sulfates at a concentration of 1 mg/mL

The results in Figure 5 show that all samples with a DS \leq 0.58 inhibited the proliferation of cells. Starting with a DS \geq 0.66, a stimulation of FGF-2-induced proliferation was observed. With an increasing degree of sulfation, the mitogenic effect of CSs was increased, too. For the derivative with the highest degree of sulfation CS 1.94, a proliferation of 160% was detected, compared with 10 ng/mL FGF-2 used here as a control.



Figure 5. Comparison of fibroblast growth after the addition of FGF-2 (10 ng/mL) and 1 mg/mL cellulose sulfates (CS) with different degrees of sulfation (CS 0.37-CS 1.94).

The proliferation was determined via the content of DNA (average \pm standard deviation, n = 5).

In general, the proliferation correlates with the binding affinity of the growth factor, which means that an increase in the sulfation degree of CSs results in a stronger FGF-2 induced mitogenic activity. CSs with a DS \leq 1.57 that showed no binding of FGF-2 could also lead to an increased proliferation. The reason for this could be based on the high affinity of FGF-2 to heparin, so that those derivatives in the competitive binding assav (Figure 4) could not bind the growth factor. Without the heparin however, the binding of FGF-2 was possible in the cellular assay, which is then visible by the increased FGF-2 activity expressed by cell growth. Similar to these results, Kunou et al. could prove an increase of FGF-1 induced proliferation through dextran sulfate with a DS of about 1.0 (26).

Dependence of mitogenic activity of cellulose sulfates on the concentration in comparison to heparin

In addition to degree of sulfation and position of the sulfate groups, the concentration of heparinanalogous CSs also influences the activity of growth factors. In studies of other authors, concentrations of heparin-analogous substances in a range of less than 1 μ g/mL could already increase FGF-2 activity toward cells. Because of this, the analysis of the mitogenic activity of cellulose sulfates was performed in the concentration range from 1 μ g/mL to 1 mg/mL. The mitogenic activity was compared with the fibroblast growth in the presence of 10 ng/mL FGF-2 as a control group.

In the following experiments, CSs with a DS of \geq 0.92 and heparin were used. In Figure 6A and 6B, a strong dependence of the proliferation on the concentration of the samples is visible. At a concentration of 1 mg/mL, with the exception of CS-0.92, all cellulose sulfates showed a proliferation of

3T3 fibroblast cells, that was comparable to that of heparin. In a concentration range of 1 – 500 μ g/mL, a stepwise concentration dependent increase in proliferation of 3T3 fibroblasts was visible from low- to high-sulfated derivatives.



Figure 6. Comparison of proliferation in dependence of the concentration and degree of sulfation of cellulose sulfates (CS) and heparin. 3T3-L1 fibroblasts were incubated with 10 ng/mL FGF-2 and 1 µg/mL bis 1 mg/mL of the derivatives for 48 hours. The proliferation was determined through the DNA content with Pico green.

(A) Heparin and medium sulfated CS, (B) high sulfated CS (average \pm standard deviation, n = 5).

The results of these investigations show that with an increasing degree of sulfation, lesser amounts of CS were needed to enhance the FGF-2 induced proliferation. It is important to note that highly sulfated celluloses are needed in significantly lower concentrations than heparin to promote cell proliferation. Since these CSs also show an increased sulfation in C2 position of AGU, this correlates with the increased affinity of FGF-2 to these CSs. Heparin-binding growth factors like FGF-2 show special, often slit-like domains that are rich in basic and partly also hydrophobic amino acids and connect to the charged chains of helically ordered heparin as stretched chains (27). One cause for the comparable or superior activity of these CSs (compared to heparin) could be their high degree of sulfation. The heparin that was used in this experiment possesses a DS of 1.3, which is below the DS of some CS used in these studies (22). Nevertheless, an important role for the interactions between heparin and FGF-2 and the interaction with the FGF-2 receptor on the cell surface is related to the carboxy group in C6 position of the iduronic acid of heparin (20). In the case of a relatively high sulfation in C2 and C6 position, the β -(1 \rightarrow 4) glycosidic bond of AGU in cellulose should lead to a homogeneous charge density on both sides of the chain, which could facilitate the binding of FGF-2 to CSs through Coulomb interactions and hydrogen bonds from remaining hydroxyl group of AGU.

The increase in the mitogenic activity of FGF-2 in combination with CS can be traced back to two causes. Growth factors like FGF-2 have a relatively short half-life period also in vitro because of the rapid proteolytic fission of the protein through proteinases released by cells. In vitro experiments have shown that the stability of FGF-2 against proteinases can be increased in the presence of heparin or highly sulfated celluloses because of the interaction between polysaccharide and growth factor (28). On the other hand, the high DS of CSs, which is comparable to heparin, could lead to the formation of a FGF-2 - CS - FGF receptor-complex on the surface of 3T3 fibroblasts, which leads to an activation of the mitogen-activated protein kinases pathway (MAPK/ ERK). The research of other authors has shown that especially high sulfation of heparin in C6 position of the glucosamine monomer plays an important role for this effect (29). Because of this and the β -(1 \rightarrow 4) glycosidic bond, a higher sulfation in the C6 and C2 position of the AGU of cellulose could be advantageous, which in the end could lead to an enhanced growth of cells in the presence of FGF-2 and higher sulfated celluloses.

Conclusion

In this study CSs were synthesized, which show a bioactivity that rivals or even surpasses that of heparin. It is obvious that a complete sulfation in C6 position and a higher sulfation in C2 position is very important for the biological activity of CS both in anti-coagulation but also promoting mitogenic activity of the heparin-binding growth factor FGF-2. This was evident by the inactivation of thrombin, which is due to a specific interaction with AT III that plays a significant role in the inhibition of blood coagulation. Although the activity of higher sulfated celluloses suggests a medical application, a direct systemic use by intravenous injection to inhibit blood clotting like heparin is not advisable due to reasons of product safety. On the other hand, immobilization of CSs on the surface of medical devices like catheters and tube systems, which have contact to blood could be an interesting alternative to heparin to increase the blood compatibility of biomaterials (30). Aside from the described inhibition of blood coagulation, CS can also bind different growth factors and influence their activity. The here described stimulating effect on the growth factor FGF-2, which has mitogenic and angiogenic effects on cells and tissues, and recent studies showing that CS have modulating effects on other growth factors like the bone morphogenic protein (BMP-2), suggest applications as bioactive coatings on surfaces of implants in the field of tissue engineering (31, 32).

Acknowledgement

I sincerely thank Mr. Wolfgang Wagenknecht as a former colleague from the Fraunhofer Institute for Polymer Science in Potsdam-Golm for the excellent and pleasant cooperation during the regioselective synthesis of cellulose derivatives and the determination of blood compatibility. This work was partially supported by grants from:

Deutsche Forschungsgemeinschaft Gr1290/12-1 and Gr1290/13-1.

References

- 1. Varki A, Cummings RD, Esko JD, Freeze HH, Stanley P, Bertozzi CR, Hart GW, Etzler ME: Essentials of Glycobiology. New York (NY): Cold Spring Harbor Laboratory Press; 2009. [PubMed]
- 2. Kresse H, Schönherr EJ. Cell Phys 2001;189:266-74. [CrossRef] [PubMed]
- 3. Garg HG, Linhardt RG, Hales CA. Chemistry and Biology of Heparin and Heparan Sulfate. San Diego (CA): Elsevier; 2005.
- 4. Capila I, Linhardt RJ. Angewandte Chemie 2002; 114: 426-50. [CrossRef]
- Bishop JR, Schuksz M, Esko JD. Nature 2007;446: 1030-7. [CrossRef] [PubMed]
- 6. Zehnder JL, Galli SJ. Nature 1999;400:714-5. [CrossRef] [PubMed]
- Lassila R, Lindstedt K, Kovanen P. Thromb. Haemostasis 1993;69:707. [CrossRef]
- Lindstedt KA, Kokkonen JO, Kovanen PT. J Lipid Res 1992;33(1):65-75. [PubMed]
- 9. Sakaguchi K, Yanagishita M, Takeuchi Y, Aurbach GD. J Biol Chem 1991;266:7270-8.
- 10. Schlessinger J, Plotnikov AN, Ibrahimi OA, Eliseenkova AV, Yeh BK, Yayon A, et al. Molec Cell 2000;6:743-50. [CrossRef] [PubMed]
- 11. Xingbin A, Do AT, Lozynska O, Gullberg MK, Lindahl U, Emerson Jr CP. J Cell Biol 2003;162:341-51. [CrossRef] [PubMed]
- Huang XJ, Guduru D, Xu ZK, Vienken J, Groth T. Macromol. Biosci 2011;11:131-40.
 [CrossRef] [PubMed]
- Street JT, McGrath M, O'Regan K, Wakai A, McGuinness A, Redmond HP. Clin Orthop Relat Res 2000;381:278-89. [CrossRef] [PubMed]
- 14. Nelson-Piercy C. Scand J Rheumatol 1998; 107: (Suppl)68-71. [CrossRef] [PubMed]
- Guerrini M, Beccati D, Shriver Z. Nature Biotechnology 2008;26:669-75. [CrossRef] [PubMed]
- Klemm D, Philipp B, Heinze T, Heinze U, Wagenknecht W. Comprehensive cellulose chemistry. 1st ed. Vol. 2. Wiley (DE): Weinheim; 1998. [CrossRef]

- 17. Heinze T, Koschella A. Macromol Symp 2005; 223: 13-29. [CrossRef]
- 18. Philipp B, Wagenknecht W. 1983;17:443-59. [CrossRef]
- 19. Groth B, Wagenknecht W. Biomaterials 2001;22: 2719-29. [CrossRef]
- 20. Casu B. Carbohydrates in Europe 1994;11:18-21. [CrossRef]
- 21. Razi N, Lindahl U. J Biol Chem 1995;270:11267-75. [CrossRef] [PubMed]
- Peschel D, Zhang K, Aggarwal N, Brendler E, Fischer S, Groth T. Acta Biomaterialia 2010;6:2116-25. [CrossRef] [PubMed]
- 23. Leali D, Belleri M, Urbinati C, Coltrini D, Oreste P, Zoppetti G, et al. J Biol Chem 2001;276:37900-8.
- 24. Faham S, Hileman RE, Fromm JR, Linhardt RJ, Rees DC. Science 1996;271:1116-20. [CrossRef] [PubMed]
- Hatanaka K, Ohtsuki T, Kunou M. Chem Lett 1994; 1407-10. [CrossRef]
- 26. Kunou M; Hatanaka K. Carbohyd. Polym 1995;28: 107-12. [CrossRef]
- 27. Nimni ME. Biomaterials 1997;18:1201-25. [CrossRef]
- Weltrowski A, da Silva Almeida ML, Peschel D, Zhang K, Fischer S, Groth T, Macromol Biosci 2012;12:740-50. [CrossRef] [PubMed]
- Rusnati M, Coltrini D, Caccia P, Dellera P, Zoppetti G, Oreste P, et al. Biochem Biophys Res Comm 1994; 203:450-8. [CrossRef] [PubMed]
- Gericke M, Doliska A, Stana J, Liebert T, Heinze T, Stana-Kleinschek K. Macromol Biosci 2011;11:549-56. [CrossRef] [PubMed]
- Peschel D. Zhang K, Fischer S, Groth T. Acta Biomaterialia 2012;8:183-93. [CrossRef] [PubMed]
- Aggarwal N, Altgärde N, Svedhem S, Zhang K, Fischer S, Groth T. Langmuir 2013;29:13853-64.
 [CrossRef] [PubMed]

Originalni rad

UDC: 615.273:676.164 doi:10.5633/amm.2020.0308

RAZVOJ BIOAKTIVNIH SULFATA CELULOZE ZA BIOMEDICINSKE SVRHE

Thomas Groth^{1,2}, Christian Willems¹, Kai Zhang³, Steffen Fischer⁴

¹Univerezitet Martin-Luther Halle-Wittenberg, Departman za biomedicinske materijale, Institut za farmaciju, Hale (Saale), Nemačka

²Interdiscipl^íinarni centar za nauku o materijalima, Univerezitet Martin-Luther Halle-Wittenberg, Hale (Saale), Nemačka

³Univerzitet Georg-August u Getingenu, Tehnologija drveta i hemija drveta, Gitnengen (Göttingen), Nemačka ⁴Tehnički univerzitet u Drezdenu, Institut za hemiju biljaka i drveta, Drezden (Dresden), Nemačka

Kontakt: Thomas Groth Heinrich-Damerow-Strasse 4, 06120 Halle, Saale, Nemačka E-mail: thomas.groth@pharmazie.uni-halle.de

Celuloza je jedan od najzastupljenijih biomolekula na zemlji. Hemijski derivati celuloze našli su široku primenu i koriste se u industrijske i biotehnološke svrhe. Sulfati celuloze (SC) predstavljaju klasu derivata rastvorljivih u vodi, koji se primenjuju u industriji, ali još uvek ne i u medicini. U ovom radu, ispitivani su derivati različitog stepena sulfatacije anhidroglukozne jedinice celuloze u cilju postizanja antikoagulantnih i modulacionih efekata faktora rasta sa heparin vezujućim domenima, poput faktora rasta fibroblasta 2 (eng. *fibroblast growth factor* 2 – FGF-2). Rezultati su pokazali da SC višeg stepena sulfatacije imaju antikoagulantnu aktivnost, koja se može porediti sa aktivnošću heparina sa udruženim dejstvom na antitrombin III, koji inhibira aktivnost trombina i faktora Xa, što SC čini interesantnim za antikoagulantna oblaganja medicinskih uređaja. Štaviše, studije su pokazale da SC sa stepenom sulfatacije uporedivim heparinu imaju promovišuću aktivnost na mitogeni efekat

Acta Medica Medianae 2020;59(3):56-67.

Ključne reči: sulfati celuloze, stepen sulfatacije, antikoagulacija, trombin, faktori rasta, FGF-2

This work is licensed under a Creative Commons Attribution 4.0 International (CC BY 4.0) Licence